opposite<br>charges<br>attract

like charges<br>repel

 $\blacksquare$ 







# Chapter 21 **Electric Charge**



**Positive and negative charges** 





# **21 Electric Charge**



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For Your Information (FYI)

Physics of electromagnetism -> combination of electric and magnetic Begin with electrical phenomena; first step is to discuss the<br>nature of electric charge and electric force.<br>Every object contains a vant amount of electric charge,

# **21-2** Electric Charge

- There are two kinds of charge: **positive** and **negative**.
- An object with equal amounts of the two kinds of charge is electrically **neutral**, whereas one with an imbalance is electrically **charged**.

 $\blacksquare$  Charges with the same electrical sign repel each other, and charges with opposite electrical signs attract each other.



Plastic

- (a) The two glass rods were each rubbed with a silk cloth (positively charged) and one was suspended by thread. When they are close to each other, they repel each other.
- (b) The plastic rod was rubbed with fur (negatively charged). When brought close to the glass rod, the rods attract each other.

*Atoms consist of positively charged protons, negatively charged electrons, and electrically neutral neutrons. The protons and neutrons are packed tightly together in a central nucleus*.<br>Copyright © 2014 John Wiley & Sons, Inc. All rights reserved.



Fig. 21-2  $(a)$  Two charged rods of the same sign repel each other.  $(b)$  Two charged rods of opposite signs attract each other. Plus signs indicate a positive net charge, and minus signs indicate a negative net charge.

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### **21-3** Conductors and Insulators FYI

# **Electric Charges in Solids**

- *When atoms of a conductor come together to form the solid, some of their outermost (and so most loosely held) electrons become free, leaving behind positively charged atoms.*
- In macroscopic solids, nuclei often arrange themselves into a regular pattern called a "Crystal Lattice".



Electrons move around this lattice.

**Red Circles = Static Positive Charge (Nuclei)**

Depending on "How They Move" the solid can be classified by its "Electrical Properties" as an Insulator or a Conductor. We call the mobile electrons as conduction electrons.

 **Conductors** are materials through which charge can move **freely**; examples include metals (such as copper in common lamp wire), the human body, and tap water.

Forming a "Sea" of electrons. This is why metals conduct electricity.

Under + - Applied Potential Difference V  $F$ **Resistance** & current **Blue Background = Mobile Electrons**

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### **21-3** Conductors and Insulators FYI

### **Materials classified based on their ability to move charge (Conductance)**

- **Nonconductors** also called **insulators** are materials through which charge cannot move freely; examples include wood, rubber, plastic, glass, and chemically pure water.
- There are few (if any) free electrons in a nonconductor.

In an **Insulator**, each electron cloud is tightly bound to the protons in a nucleus. But the electron cloud can "Distort" locally.



but polarization & dipole moment

 **Semiconductors** are materials that are intermediate between conductors and insulators; examples include silicon and germanium in computer chips. **Superconductors** are materials that are perfect conductors, allowing charge to move without any hindrance (no resistance!). on off states

http://www.phys.lsu.edu/~jdowling/PHYS21132-SP15/lectures/jndex.html

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# **21-3 Conductors and Insulators**





Fig. 21-4 A neutral copper rod is electrically isolated from its surroundings by being suspended on a nonconducting thread. Either end of the copper rod will be attracted by a charged rod. Here, conduction electrons in the copper rod are repelled to the far end of that rod by the negative charge on the plastic rod. Then that negative charge attracts the remaining positive charge on the near end of the copper rod, rotating the copper rod to bring that near end closer to the plastic rod.

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### **Coulomb's Law**  $\vec{F} = k \frac{q_1 q_2}{r^2} \hat{r}$ (Coulomb's law),

- The force of repulsion/attraction due to the charge properties of objects is called an **electrostatic force.** Why static? We study the snapshot of the system.
- Coulomb's law describes the **electrostatic force** (or electric force) between two charged particles.

### • The magnitude of the electric forces between charged objects is measured by "Torsion Balance" which was invented by Charles Augustin De Coulomb. <http://www.dnatube.com/video/11874/Application-Of-Coulombs-Torsion-Balance>

- $\bullet$  If the particles have charges  $\boldsymbol{q}_2$  and  $\boldsymbol{q}_7$ , are separated by distance  $r$ , and are *at rest (or moving only slowly)* relative to each other, then the  $F =$   $\frac{1}{\sqrt{1 - x^2}}$ magnitude of the force acting on each due to the other is given by
- Note that Newton's Third Law says  $|F_{12}| = |F_{21}|$  $F_{12} = \frac{1}{4\pi\epsilon_0} \frac{|q_1||q_2|}{r^2}$

where 
$$
\varepsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{N} \cdot \text{m}^2
$$
 is the constant for permittivity of free space.

- Response of the medium to the external  $\vec{E}$
- If not air/vacuum: *ε<sup>0</sup> → ε*
- $\bullet$  The ratio  $1/4\pi\varepsilon_o$  is often replaced with the electrostatic constant (or Coulomb constant) *k=8.99×10<sup>9</sup> N.m<sup>2</sup> /C<sup>2</sup>* .

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Just calculate magnitude(s) of forces!  $q_{9}$ Copyright © 2014 John Wiley & Sons, Inc. All rights reserved  $\vec{F} = |\vec{F}| \hat{r}$ 

First make analysis of magnitudes of the forces then introduce (unit) vector

Charles-Augustin De Coulomb (1736–1806)







uniformly over the (external) surface. Now, first shell theorem works!

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- The coulomb unit is derived from SI unit ampere for electric current i.
- Current is the rate  $\Delta q/\Delta t$  at which charge moves past a point or through a region

$$
i = \frac{dq}{dt}
$$
 (electric current),

in which i is the current (in amperes) and dq (in coulombs) is the amount of charge moving past a point or through a region in time dt (in seconds).

• Therefore,  $1 C = (1 A)(1 s)$ .



### **Example, The net force due to two other particles:**



**Fig. 21-8 (***a)* Two charged particles of charges  $q_1$  and  $q_2$  *are fixed in* place on an *x axis.* (*b) The free-body* diagram for particle 1, showing the electrostatic force on it from particle 2.

 $\vec{F}_{1,net} = |\vec{F}_{1,net}| \hat{r}$ 

(a) Figure 21-8a shows two positively charged particles fixed in place on an x axis. The charges are  $q_1 = 1.60 \times 10^{-19}$  C and  $q_2 =$  $3.20 \times 10^{-19}$  C, and the particle separation is  $R = 0.0200$  m. What are the magnitude and direction of the electrostatic force  $\vec{F}_{12}$  on particle 1 from particle 2?

$$
F_{12} = \frac{1}{4\pi\epsilon_0} \frac{|q_1||q_2|}{R^2}
$$
  
= (8.99 × 10<sup>9</sup> N·m<sup>2</sup>/C<sup>2</sup>)  
×  $\frac{(1.60 × 10^{-19} C)(3.20 × 10^{-19} C)}{(0.0200 m)^2}$   
= 1.15 × 10<sup>-24</sup> N. First magnitude then direction

Thus, force  $\vec{F}_{12}$  in unit-vector notation is

$$
\vec{F}_{12} = -(1.15 \times 10^{-24} \,\text{N})\hat{\mathbf{i}}.\qquad\qquad \text{(Answer)}
$$
\n
$$
|\vec{F}_{12}|(-\hat{\mathbf{i}})
$$

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### **Example, The net force due to two other particles, cont.:**

(b) Figure 21-8 $c$  is identical to Fig. 21-8 $a$  except that particle  $\vec{F}_{1,net} = \vec{F}_{12} + \vec{F}_{13}$   $\vec{F}_{12}$   $\sqrt{ }$ 3 now lies on the  $x$  axis between particles 1 and 2. Particle 3 has charge  $q_3 = -3.20 \times 10^{-19}$  C and is at a distance  $\frac{3}{4}R$  from particle 1. What is the net electrostatic force  $\vec{F}_{1,\text{net}}$  on particle  $\eta_{13}^j = \frac{1}{4\pi\epsilon_0} \frac{|q_1||q_3|}{(\frac{3}{4}R)^2}$ 1 due to particles 2 and 3?



**Fig. 21-8** (c) Particle 3 included. (*d) Free-body diagram* for particle 1.

Screening effect! Superposition

 $principle \rightarrow NO$ Screening

 $\times \frac{(1.60 \times 10^{-19} \text{ C})(3.20 \times 10^{-19} \text{ C})}{(\frac{3}{4})^2 (0.0200 \text{ m})^2}$  $= 2.05 \times 10^{-24}$  N.

+x-direction

We can also write  $\vec{F}_{13}$  in unit-vector notation:

 $= (8.99 \times 10^{9} \,\mathrm{N} \cdot \mathrm{m}^{2}/\mathrm{C}^{2})$ 

 $\vec{F}_{13} = (2.05 \times 10^{-24} \,\text{N})\hat{\text{i}}$ .

The net force  $\vec{F}_{1,net}$  on particle 1 is the vector sum of  $\vec{F}_{12}$ and  $\vec{F}_{13}$ ; that is, from Eq. 21-7, we can write the net force  $\vec{F}_{1,\text{net}}$  on particle 1 in unit-vector notation as

$$
\vec{F}_{1,net} = \vec{F}_{12} + \vec{F}_{13}
$$
  
= -(1.15 × 10<sup>-24</sup> N) $\hat{i}$  + (2.05 × 10<sup>-24</sup> N) $\hat{i}$   
= (9.00 × 10<sup>-25</sup> N) $\hat{i}$ . (Answer)

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### **Example, The net force due to two other particles, cont.:**

(c) Figure 21-8e is identical to Fig. 21-8a except that particle  $\vec{F}_{1,net} = \vec{F}_{12} + \vec{F}_{14}$ 4 is now included. It has charge  $q_4 = -3.20 \times 10^{-19}$  C, is at a distance  $\frac{3}{4}R$  from particle 1, and lies on a line that makes an angle  $\theta = 60^{\circ}$  with the x axis. What is the net electrostatic  $F_{\!\!14}^{\!\!\frac{1}{4}} = \frac{1}{4\pi\varepsilon_0} \frac{|q_1||q_4|}{(\frac{3}{4}R)^2}$ force  $\vec{F}_{1,\text{net}}$  on particle 1 due to particles 2 and 4? This is the third  $\overrightarrow{F}_{12}$   $\sqrt{ }$ =  $(8.99 \times 10^{9} \,\mathrm{N \cdot m^{2}/C^{2}})$  $\times \frac{(1.60 \times 10^{-19} \text{ C})(3.20 \times 10^{-19} \text{ C})}{(\frac{3}{4})^2 (0.0200 \text{ m})^2}$ arrangement.  $\vec{F}_{14} \Rightarrow x \& y$  $= 2.05 \times 10^{-24}$  N.  $components$  $(e)$  $\vec{F}_{14} = (F_{14} \cos \theta) \hat{i} + (F_{14} \sin \theta) \hat{j}.$ This is still the Substituting  $2.05 \times 10^{-24}$  N for  $F_{14}$  and 60° for  $\theta$ , this becomes particle of interest.  $\vec{F}_{14} = (1.025 \times 10^{-24} \text{ N})\hat{i} + (1.775 \times 10^{-24} \text{ N})\hat{j}.$ Then we sum:  $F_{14,x}$   $F_{14,y}$ It is pulled toward  $\vec{F}_{1,net} = \vec{F}_{12} + \vec{F}_{14}$ particle 4.  $= -(1.15 \times 10^{-24} \text{ N})\hat{i}$ +  $(1.025 \times 10^{-24} \text{ N})\hat{i} + (1.775 \times 10^{-24} \text{ N})\hat{j}$  $-$  It is pushed away  $\approx (-1.25 \times 10^{-25} \text{N})\hat{i} + (1.78 \times 10^{-24} \text{N})\hat{j}.$ from particle 2. in vectorial form (Answer)

**Fig. 21-8** (*e)* Particle 4 included. (*f )* Freebody diagram for particle 1.

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### **Example, The net force due to two other particles, cont.:**



**Fig. 21-8** (*e)* Particle 4 included. (*f )*  Freebody diagram for particle 1.

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 $F_{1,net x} = F_{12x} + F_{14x} = F_{12} + F_{14} \cos 60^{\circ}$ 

 $= (2.05 \times 10^{-24} \text{ N})(\sin 60^{\circ})$ 

 $F_{\text{1 net}} = \sqrt{F_{\text{1 net }x}^2 + F_{\text{1 net }y}^2} = 1.78 \times 10^{-24} \text{ N}.$ 

However, this is an unreasonable result because  $\vec{F}_{1,net}$  must have a direction between the directions of  $\vec{F}_{12}$  and  $\vec{F}_{14}$ . To

 $= -1.25 \times 10^{-25}$  N.

 $= 1.78 \times 10^{-24}$  N.

To find the direction of  $\vec{F}_{1,net}$ , we take

correct  $\theta$ , we add 180 $^{\circ}$ , obtaining

 $120^\circ$ 

rotation

Positive Angle- rotates counter-clockwise  $(CCW)$ Negative Angle- rotates clockwise (CW)

 $\theta = \tan^{-1} \frac{F_{1,\text{net},y}}{F_{1,\text{net},x}} = -86.0^{\circ}.$  CW (from -x)

 $-86.0^{\circ} + 180^{\circ} = 94.0^{\circ}$  CCW  $\left(\frac{f_{TQ}}{f_{TS}Q} + x\right)$ <br>Angle describes the **amount** and **direction** of

 $-210^\circ$ 

 $= -1.15 \times 10^{-24} \text{ N} + (2.05 \times 10^{-24} \text{ N})(\cos 60^{\circ})$ 



### **Example, Equilibrium of two forces:**

Figure 21-9a shows two particles fixed in place: a particle of charge  $q_1 = +8q$  at the origin and a particle of charge  $q_2 = -2q^2$ at  $x = L$ . At what point (other than infinitely far away) can a proton be placed so that it is in *equilibrium* (the net force on it is zero)? Is that equilibrium *stable* or *unstable*? (That is, if the proton is displaced, do the forces drive it back to the point of equilibrium or drive it farther away?)





Χ

 $\overline{0}$ 

**Example, Equilibrium of two forces, cont.:**

Instant stability

The equilibrium at  $x=2L$  is unstable that is, if the proton is displaced leftward from point *R,*  then  $F_1$  and  $F_2$  both increase but  $F_2$  increases more (because  $q_2$  is closer than  $q_1$ ), and a net force will drive the proton farther leftward. If the proton is displaced rightward, both  $F<sub>1</sub>$  and *F2* decrease but *F<sup>2</sup>* decreases more, and a net force will then drive the proton farther rightward. In a stable equilibrium, if the proton is displaced slightly, it returns to the equilibrium position.

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In Fig. 21-10a, two identical, electrically isolated conducting spheres  $A$  and  $B$  are separated by a (center-to-center) distance  $\overline{a}$  that is large compared to the spheres. Sphere  $\overline{A}$  has a positive charge of  $+Q$ , and sphere B is electrically neutral. Initially, there is no electrostatic force between the spheres. (Assume that there is no induced charge on the spheres because of their large separation.)

(a) Suppose the spheres are connected for a moment by a conducting wire. The wire is thin enough so that any net charge on it is negligible. What is the electrostatic force between the spheres after the wire is removed?



**Fig. 21-10** *Two small conducting spheres A and B. (a) To start, sphere A is charged positively. (b) Negative charge is transferred from B to A through a connecting wire.* (*c) Both spheres are then charged* positively.

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## **Example, Charge Sharing:**

Shell theorem I

(1) Since the spheres are identical, connecting them means that they end up with identical charges (same sign and same amount). (2) The initial sum of the charges (including the signs of the charges) must equal the final sum of the charges.

*Reasoning*: When the spheres are wired together, the (negative) conduction electrons on B move away from one another (along the wire to positively charged A— Fig. 21- 10b.)

As B loses negative charge, it becomes positively charged, and as A gains negative charge, it becomes less positively charged. The transfer of charge stops when the charge on B has increased to Q/2 and the charge on A has decreased to Q/2, which occurs when Q/2 has shifted from B to A. & (2) charge conservation  $Q_i = Q_f$ 

The spheres, now positively charged, repel each other.

$$
F = \frac{1}{4\pi\epsilon_0} \frac{(Q/2)(Q/2)}{a^2} = \frac{1}{16\pi\epsilon_0} \left(\frac{Q}{a}\right)^2.
$$



## **Example, Charge Sharing, cont.:**

(b) Next, suppose sphere  $A$  is grounded momentarily, and then the ground connection is removed. What now is the electrostatic force between the spheres?



**Fig. 21-10** (*d) Negative charge is transferred through a grounding wire to sphere A. (e) Sphere A is then neutral.* Ground  $\rightarrow$  neutral (large enough reservoir) Creating a pathway between object and Earth's surface

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*Reasoning:* When we provide a conducting path between a charged object and the ground (which is a huge conductor), we neutralize the object.

Where sphere *A* negatively charged, the mutual repulsion between the excess electrons would cause them to move from the sphere to the ground.

However, because sphere *A* is positively charged, electrons with a total charge of *Q/2*  move from the ground up onto the sphere (Fig. 21-10*d),* leaving the sphere with a charge of 0 (Fig. 21-10*e).* Thus, there is (again) no electrostatic force between the two spheres.



# **Example, with symmetry:** Charge +q placed at center



What is the Force on central particle?

All forces cancel except from +2q! target

consider the symmetry!



without considering the symmetry: 11

$$
\vec{F}_{+q,net} = \sum_{i=1} \vec{F}_{+q,i}
$$

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### **21-5** Charge is Quantized FYI

- Electric charge is quantized (restricted to certain values). The total charge of any object is found to always be a multiple of a certain **elementary charge**, "e":
- The value of this elementary charge is one of the fundamental constants of nature, and it is the magnitude of the charge of both the proton and the electron.



 $e = 1.602 \times 10^{-19}$  C.

- Elementary particles either carry no charge, or carry a single elementary charge.
	- Since the days of Benjamin Franklin, our understanding of of the nature of electricity has changed from being a type of '*continuous fluid' (not correct!) to a collection of smaller charged particles*.
- The charge of a particle can be/written as *ne*, where *n* is a positive or negative integer and *e* is the elementary charge.
- Any positive or negative charge q that can be detected can be written as Benjamin Franklin (1705–1790)  $q = n\ell$ ,  $n = \pm 1, \pm 2, \pm 3, \ldots$
- It is possible, for example, to find a particle that has no charge at all, or a charge of +10e, or -6e, but not a particle with a charge of, say, 3.57e.
- One cannot ISOLATE FRACTIONAL CHARGE (e.g.  $-1/2$  e,  $+1/3$  e, etc.)





### **21-5** Charge is Quantized FYI



- Many descriptions of electric charge use terms that might lead you to the conclusion that charge is a substance. Phrases like: "Charge on a sphere" "Charge transferred"
	- "Charge carried on the electron"
- However, charge is a *property* of **particles,** one of many properties, such as mass.

Mass is an intrinsic property as well as charge.

# **21-5** Charge is Quantized



### **Example, Mutual Electric Repulsion in a Nucleus:**

Atomic sizes  $\rightarrow 10^{-10}$  m (Angstrom or nanometer)

The nucleus in an iron atom has a radius of about  $4.0 \times$  $10^{-15}$  m and contains 26 protons.

(a) What is the magnitude of the repulsive electrostatic force between two of the protons that are separated by  $4.0 \times 10^{-15}$  m?

### **KEY IDEA**

The protons can be treated as charged particles, so the magnitude of the electrostatic force on one from the other is given by Coulomb's law.

**Calculation:** Table 21-1 tells us that the charge of a proton is  $+e$ . Thus, Eq. 21-4 gives us

$$
F = \frac{1}{4\pi\epsilon_0} \frac{e^2}{r^2}
$$
  
= 
$$
\frac{(8.99 \times 10^9 \,\mathrm{N \cdot m^2/C^2})(1.602 \times 10^{-19} \,\mathrm{C})^2}{(4.0 \times 10^{-15} \,\mathrm{m})^2}
$$
  
= 14 N.  $\leftarrow$  Small or NOT? (Answer)

**No explosion:** This is a small force to be acting on a macroscopic object like a cantaloupe, but an enormous force to be

acting on a proton. Such forces should explode the nucleus of any element but hydrogen (which has only one proton in its nucleus). However, they don't, not even in nuclei with a great many protons. Therefore, there must be some enormous attractive force to counter this enormous repulsive

electrostatic force.<br>
(b) What is the magnitude of the gravitational force between those same two protons?

### **KEY IDEA**

Because the protons are particles, the magnitude of the gravitational force on one from the other is given by Newton's equation for the gravitational force (Eq. 21-2).

**Calculation:** With  $m_p$  (= 1.67  $\times$  10<sup>-27</sup> kg) representing the mass of a proton, Eq. 21-2 gives us

$$
F = G \frac{m_{\rm p}^2}{r^2}
$$
 not effective in atomic sizes  
= 
$$
\frac{(6.67 \times 10^{-11} \,\mathrm{N \cdot m^2/kg^2})(1.67 \times 10^{-27} \,\mathrm{kg})^2}{(4.0 \times 10^{-15} \,\mathrm{m})^2}
$$
  
= 1.2 × 10<sup>-35</sup> N. very small (Answer)

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# **21-6** Charge is Conserved



### **CHECKPOINT 4**

Initially, sphere A has a charge of  $-50e$  and sphere B has a charge of  $+20e$ . The spheres are made of conducting material and are identical in size. If the spheres then touch, what is the regulting charge on sphere  $A$ ?

Charges are located at surface



isolated system is always conserved.

• The net electric charge of any



Long distance No electrostatic force No induction



B

 $\overline{+}$ 

+

+

A 、「」<br>A 、「」

 $-50e$   $+20e$ 

i<br>F :<br>E י<br>⊑י

**F** 

- - -<br>3

**F** 

ر در<br>المال

/ اليه<br>ا

No contact. But induction: charges are relocated!





1) What must be the distance between point charge  $Q_1 = 26.0 \mu C$ and point charge  $Q_2$  = -47.0  $\mu$ C for the electrostatic force between them to have a magnitude of 5.70 N? should be in SI unit



2) In Figure, three charged particles lie on an *x* axis. Particles 1 and 2 are fixed in place. Particle 3 is free to move, but the net electrostatic force on *it* from particles 1 and 2 happens to be zero. If  $\underline{L}_{23} = \underline{L}_{12}$ , what is the ratio  $q_1/q_2$ ?

equal distances<br>
3 charged particles }<br>
12 x are fixed<br>
3 is free to move<br>
F<br>
1-0  $\overrightarrow{f}_{3,net} = \sum_{i=1}^{2} \overrightarrow{f}_{3i} = \overrightarrow{f}_{3i} + \overrightarrow{f}_{32} = k \frac{|q_3||q_1|}{d_{31}^2} + k \frac{|q_3||q_2|}{d_{32}^2} = 0$ <br> $\rightarrow \frac{q_1}{l^2} = 0 \rightarrow \frac{q_1}{q_2} = -4$  why minus sign?

 $L_{23}$ 

 $L_{23} = L_{12}$ 





3) In Figure shown, what are i) horizontal components of the net electrostatic force on the charged particle in the lower left corner of the system ii) vertical component of that.

ii) vertical components of that

\n
$$
f(z) = k \frac{|z_0|}{a^2} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}
$$
\n
$$
= k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}
$$
\n
$$
= k \frac{2q^2}{a^2}, \overline{F}_{13} = k \frac{(4q^2)}{a^2}, \overline{F}_{14} = k \frac{(4q^2)}{a^2}, \overline{F}_{15} = k \frac{(4q^2)}{a^2}
$$
\n
$$
= k \frac{2q^2}{a^2}, \overline{F}_{14} = k \frac{(4q^2)}{a^2}, \overline{F}_{15} = k \frac{(4q^2)}{a^2}, \overline{F}_{16} = k \frac{(4q^2)}{a^2}, \overline{F}_{17} = k \frac{(4q^2)}{a^2}, \overline{F}_{18} = k \frac{(4q^2)}{a^2}, \overline{F}_{19} = k \frac{(4q^2)}{a^2}, \overline{F}_{10} = k \frac{(4q^2)}{a^2}, \overline{F}_{11} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a^2}, \overline{F}_{13} = k \frac{(4q^2)}{a^2}, \overline{F}_{14} = k \frac{(4q^2)}{a^2}, \overline{F}_{15} = k \frac{(4q^2)}{a^2}, \overline{F}_{16} = k \frac{(4q^2)}{a^2}, \overline{F}_{17} = k \frac{(4q^2)}{a^2}, \overline{F}_{18} = k \frac{(4q^2)}{a^2}, \overline{F}_{19} = k \frac{(4q^2)}{a^2}, \overline{F}_{10} = k \frac{(4q^2)}{a^2}, \overline{F}_{11} = k \frac{(4q^2)}{a^2}, \overline{F}_{12} = k \frac{(4q^2)}{a
$$



i)  $F_{1,net,n} = F_{12,2} + F_{13,n} + F_{14,n} = k \frac{q^2}{q^2} (4 + \frac{\sqrt{2}}{2}) = 8.99 \times 10^9 (10 \times 10^{-3})^2 (4 + \frac{\sqrt{2}}{2})$ <br>
ii)  $F_{1,net,y} = F_{12,y} + F_{13,y} + F_{14,y} = k \frac{q^2}{q^2} (-2 + \frac{\sqrt{2}}{2}) = 0 \sqrt{-0.046} N$  $F_{1,net} = k \frac{q^2}{q^2} \left[ (4 + \frac{\sqrt{2}}{2}) t^2 + (-2 + \frac{\sqrt{2}}{2}) t^2 \right]$  $F_{1,net} = 0.17 \sqrt{2} + 0.046 (-3)$  $tan \theta = -0.046$ 



4) Two identical conducting spheres, fixed in place, attract each other with an electrostatic force of 0.108 N when their center-tocenter separation is 50.0 cm. The spheres are then connected by a thin conducting wire. When the wire is removed, the spheres repel each other with an electrostatic force of 0.0360 N. Of the initial charges on the spheres, with a positive net charge, what  $\overline{a}$  the meantime shower amount of the small  $\overline{a}$  $\mathcal{L}_{\text{loc}}$  $\overline{a}$ .

a. The negative charge on one of them?   
\n
$$
a^{\frac{1}{2}}
$$
 and b. the positive charge on the other?  
\n $a^{\frac{1}{2}} = \frac{9!}{2} \times \frac{9!}{2!} \times \frac{10!}{2!} \times 10! \times 1$ 



Choose as 92 15 positive.  $F = k \frac{|q_1| |q_2|}{\rightarrow} -0.108N (0.5m)^2 - q_1 q_2$ Two, identical andrealing  $F = k \frac{|9|1|92|}{d^2} \rightarrow \frac{-0.106N(0.514)}{(8.99 \times 10^7 N m^2/c)}$ <br> $9.92 = -3.00 \times 10^{-12} c$ <br> $ln i \frac{1}{10^{1/2}}$  total cheral ( sphere  $9.92 = -3.00 \times 10^{-6}$ <br>chage balance Initial total change is -First attact each other  $2)$ shared  $(i) F = 00.108N$  $F_{=}+k\left(\frac{q_{1}+q_{2}}{2}\right)\left(\frac{q_{1}+q_{2}}{2}\right)$  $d = 50.0 cm$ - Then connected by a  $\frac{1}{2}$   $\frac{1}{2}$  conducting wire  $0.36$   $N/9$  (0.5 $m$  $\sqrt{2}$   $\sqrt{2}$   $\sqrt{2}$  $\mathcal{L}$  is a matrix  $\mathcal{L}$ Balance<br>-Whe is removed Repulsion 9,  $91.92 = 2.00 \times 10^{-6} C$  Two egns<br>(2)  $F = 40.0360 \text{ N}$  $(2) F = 70.0360N$ Assumption: Positive net  $91$ <br> $91$ <br> $92 - 206$ <br> $96 - 300 \times 10^{-12} = 0$ <br> $91 = -6 \pm \sqrt{6^{2}-482}$ <br> $91 = -6 \pm \sqrt{6^{2}-482}$ choy  $\overbrace{C}$  sign: 9=3.00 xide cd 92=-1.00 x156C  $9227$  $9,5$  $-3.00 \times 10^{-6}C$ <br> $-3.00 \times 10^{-6}C$ <br> $-1.00 \times 10^{-6}C$ <br> $= 3.00 \times 10^{-6}C$  $ax^2 + bx + c = 0$ <br>Quadratic equation:  $x_{12} = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$  [such that 91.192.70]  $ax^2 + bx + c = 0$ September 23, 2021 PHY102 Physics II © Dr.Cem Özdoğan 28



5) Two particles are fixed on an *x* axis. Particle 1 of charge 50 µC is located at  $x = -2.0$  cm; particle 2 of charge *Q* is located at  $x = 3.0$ cm. Particle 3 of charge magnitude 20 µC is released from rest on the *y* axis at  $y = 2.0$  cm. What is the value of *Q* if the initial acceleration of particle 3 is in the positive direction of (a) the *x*  axis and (b) the *y* axis (solution is not given)?

x-axis. no net force on the y-axis  $\int$  force on the v-axis  $\mathcal{P}_{\mathsf{A}}(\mathcal{G})$  $\ell$  with charge and point continuous +  $20<sub>M</sub>$ C *Q2 = -*47.0 µC for the electrostatic force between them to  $\frac{1}{2}$   $\frac{1}{2}$   $\frac{1}{2}$  $\zeta$ + **[ d= 1.39 m]**  $\frac{1-\frac{3}{32}}{9 \text{ should be negative}}$   $|F_{32}| = k \frac{120\times10^{-6}c/100}{(2\times10^{-2}m)^2+(3\times10^{-2}m)}$  $\frac{(20\times10^{-2} \text{ m})^2 + (3\times10^{-2} \text{ m})^2}{(2\times10^{-2} \text{ m})^2 + (3\times10^{-2} \text{ m})^2}$ O & Q are fined<br>3 is released from rest S showia de negative (2x10<sup>-2</sup>m)<sup>2</sup> + (3x10<sup>2</sup>m)<br>S | F<sub>31</sub> | Cos4 S = | F<sub>32</sub> | Cost x where  $\theta = \tan^{-1} \frac{3x\sqrt{6}m}{2x\sqrt{6}m} = 56.31^{\circ}$  $\frac{4}{8\times10^{-4}h^{2}}\frac{120\times10^{-6}c(150\times10^{-6})}{8\times10^{-4}h^{2}}\cos45=k\frac{120\times10^{-6}c(18)}{13\times10^{-4}m^{2}}\cos56.31^{\circ}$ 

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towards







- 6) Figure shows an arrangement of four charged particles, with angle  $\theta$ =35.0° and distance  $d$ =2.00 cm. Particle 2 has charge  $q_2$ = +8.00x10-
	- <sup>19</sup> C; particles 3 and 4 have charges  $q_3 = q_4 = -1.60 \times 10^{-19}$  C.
		- a. What is distance *D* between the origin and particle 2 if the net electrostatic force on particle 1 due to the other particles is zero? target
		- b. If particles 3 and 4 were moved closer to the *x* axis but maintained their symmetry about that axis, would the required value of *D* be greater than, less than, or the same as in part (a)?





 $F(22)$  Four charged particles  $(i)$   $D = ?$   $4f$   $[i]$ , net  $= 0$  $F_{1,2}$   $k = F_{1,2} + F_{1,3} + F_{1,2}$  Superposition  $0 = 35.0$  $d = 2.00 cm$ what about the sign of the chage 9,.  $92 = 8.00 k10$  $92 = 1.60 \times 10^{-19}$ Lets say positive  $F_{13, x} + F_{12, x}$  $\rightarrow$   $|F_{12}|=$  $k \frac{|q_1||q_2|}{\pi} = k \frac{|q_1||q_3|}{\pi} + \frac{|q_1||q_4|}{\pi}$  $(d+0)^2$  $\overline{\cos \theta}$  -1  $\leq \cos \theta \leq +1$   $|q_3|/|q_4|$  $1921$  $2|43|$   $Cor20$  $\hat{w}$  364 moved closer  $\Rightarrow \theta$  V  $(0.02m+0)^2$  $= 2*1.6x/0.2$  $\Rightarrow$  Cos  $\Theta$  increases  $cos\theta = 1$  as  $max$  $C_{15}$   $25$  $8.00\times10^{-17}C$  $\frac{1}{(0.02m+D)^2}$   $\frac{(0.02m)^2}{2}$  $=0.02m V$  $8.00 \times 10^{-2}$  $D = 10.02m + 0.02m$  $\frac{10.025}{2*1.6865}$   $\frac{10}{25}$ D decreases  $= 0.0186 m = 1.86 cm$  $\theta \downarrow \cdots cos\theta \uparrow \cdots$ 

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7) Figure (a) shows charged particles 1 and 2 that are fixed in place on an *x* axis. Particle 1 has a charge with a <u>magnitude of  $|q_1| = 8.00e$ </u>. Particle 3 of charge  $q_3 = +8.00e$  is initially on the *x* axis near particle 2. Then particle 3 is gradually moved in the positive direction of the *x* axis. As a result, the magnitude of the net electrostatic force  $F_{2,\text{net}}$ on particle 2 due to particles 1 and 3 changes. Figure (*b)* gives the *x*  component of that net force as a function of the position *x* of particle 3. The scale of the *x* axis is set by  $x_s=0.80$  m. The plot has an <u>asymptote of  $F_{2\text{net}}=1.5x10^{-25}$  N as  $x\rightarrow\infty$ </u>. As a <u>multiple of *e*</u> and including the sign, what is the charge  $q_2$  of particle 2?



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 $11 (32)$  9, 8 92 : fixed in<br> $11 - 9 + 8.00e$  place at  $x_3 = 0.04$  m  $F(x_3) = 0$ . This shows that q, has passive change.  $93 - 8.006$ particle 3 gradualy  $90.4720423$ Famet (ng) is given  $x \rightarrow \infty$  Fanetal.  $S x 10^{-25} N$ Asymptotic value of  $F_{2,net}$  as  $x_3 \rightarrow \infty$ .  $9_{2}$  = ? Take this as a maximum force the 9, 892  $k = 8.99 \times 10^{9} N m^{2}/c^{2}$ 1-5x10<sup>-25</sup>N = k 19/1/90/ 792=15x10 N d<sup>2</sup>  $e = 1.602$  N/J  $R_C$ 1.5×10<sup>25</sup> N (0.04m)<sup>2</sup><br>8.99×109N my<sub>2</sub> (8×1.602×15<sup>19</sup>C) = 2.08×10<sup>-18</sup>C T  $92 = 1.5 \times 10^{-25} N (0.04 m)^2$  $\frac{2.08 \times 10^{-18} C}{790} = 13e$  $(2)$ 



## **Electric Charge**

• The strength of a particle's electrical interaction with objects around it depends on its electric charge, which can be either positive or negative.

## **Conductors and Insulators**

• Conductors are materials in which a significant number of electrons are free to move. The charged particles in nonconductors (insulators) are not free to move.

## **Conservation of Charge**

• The net electric charge of any isolated system is always conserved.

## **Coulomb's Law**

• The magnitude of the electrical force between two charged particles is proportional to the product of their charges and inversely proportional to the square of their separation distance.

$$
F = \frac{1}{4\pi\epsilon_0} \frac{|q_1|}{}
$$

• .

$$
Eq. 21-4
$$

## **The Elementary Charge**

- Electric charge is quantized (restricted to certain values).
- *e* is the elementary charge

$$
e = 1.602 \times 10^{-19}
$$
 C. Eq. 21-12

 $|q_2|$ 



# Additional Materials

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# 21-6 Charge is Conserved

- The net electric charge of any isolated system is always conserved.
- If two charged particles undergo an annihilation process, they have equal and opposite signs of charge.

 $e^- + e^+ \rightarrow \gamma + \gamma$ 

If two charged particles appear as a result of a pair production process, they have equal and opposite signs of charge.  $\gamma \rightarrow e^- + e^+$ 



- A photograph of trails of bubbles left in a bubble chamber by an electron and a positron. The pair of particles was produced by a gamma ray that entered the chamber directly from the bottom. Being electrically neutral, the gamma ray did not generate a telltale trail of bubbles along its path, as the electron and positron did.
- **R***adioactive decay of nuclei,* in which a nucleus transforms into (becomes) a different type of nucleus.
- A uranium-238 nucleus (238U) transforms into a thorium- 234 nucleus (234Th) by emitting an *alpha particle.* An alpha particle has the same makeup as a helium-4 nucleus, it has the symbol <sup>4</sup>He. Here the net charge is 238.  $^{238}$ U  $\rightarrow$   $^{234}$ Th +  $^{4}$ He,

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